Molecular mechanisms of radiation damage in DNA: ESR and optical detection of oxidation reactions with 5-substituted uracil derivatives in frozen glasses.

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Frozen glasses form a suitable system for the study of the reactions of the indirect effect of ionizing radiation on biological macromolecules. The primary radical intermediates of the water radiolysis H', OH' and e are produced by X-irradiation at low temperatures (77 K) and can react with solute molecules upon annealing the samples at higher temperatures. With this procedure we have investigated the reactions of SO_4^{\bullet} and HPO_4^{\bullet} with several DNA constituents, especially 5-halogen-substituted uracil derivatives, in an acidic aqueous matrices (6MH2SO4, 14MH₃PO₄). The radiation chemistry of this class of compounds deserves special attention because of its radiosensitizing properties when incorporated into native DNA instead of thymine.

The products of the attack of SO_4^{\bullet} (HPO $_4^{\bullet}$) at ca. 150 K on thymine (T) and the range of 5-substituted bases (FU, ClU, BU, IU) and their nucleosides were identified by Electron Spin Resonance Spectroscopy (ESR) and by optical absorption.

Both species were found to oxidize the bases under formation

of the corresponding base cations:

$$xu + so_4^- (HPO_4^-) \rightarrow xu^+ + so_4^- (HPO_4^-)$$

Both the ESR hyperfine parameters and the optical absorption bands of the cations $\mathrm{XU}^{+\,\bullet}$ could be determined. They show a strong substituent dependence as expected from the spin density distribution.

The secondary reactions of XU+ depend on the experimental conditions and the molecule under investigation. The radical cations of free bases may deprotonate at position N1 thus forming a neutral radical

$$XU^+ \cdot \rightarrow XU \cdot + H^+$$

whereas in nucleosides in a water-rich environment the hydroxylate at C_6 is formed:

The importance of the above reactions for the radiation chemistry of in vivo DNA will be discussed.